

Flow Absorber Possibilities

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Neutrino Factory & Muon Collider Collaboration Meeting @ IIT 03/12/06 Fermi Na

03/12/06

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Flow Absorber possibilities



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Headlines:

- > Materials and fluids for Muon ionization cooling
- \succ Disadvantages of the LH₂ choice
- Review of the basic LH, Absorbers designs

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- Design philosophy for a hostile environment
- Implementing SAMPLE cryo-loop
- Cryo-system to support the LH₂ absorber tests at MTA
- Concluding comments

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Materials and fluids for Muon ionization cooling

	Material	Z	Α	$\langle Z/A \rangle$	Nuclear ^a collision length <i>λq</i> {g/cm ² }	Nuclear $\int \frac{1}{10000000000000000000000000000000000$	$\frac{dE}{dx} \frac{dE}{dx} dE$	Radiati {g/cm ²	on length ^c X ₀ } {cm}	Density $\{g/cm^3\}$ $(\{g/\ell\}$ for gas)	Liquid boiling point at 1 atm(K)	Refractive index n $((n-1)\times 10^5$ for gas)	Transverse cooling 1.2 1.2 0.8 0.6 0.4	g merit factor	$F \propto (L_R dE/dx)^2$
[H ₂ gas H ₂ liquid D ₂ He Li Be C N ₂ O ₂ F ₂	1 1 2 3 4 6 7 8 9	$\begin{array}{c} 1.00794\\ 1.00794\\ 2.0140\\ 4.002602\\ 6.941\\ 9.012182\\ 12.011\\ 14.00674\\ 15.9994\\ 18.9984032 \end{array}$	0.99212 0.99212 0.49652 0.4968 0.43221 0.44384 0.49954 0.49976 0.50002 0.47372	43.3 43.3 45.7 49.9 54.6 55.8 60.2 61.4 63.2 65.5	50.8 50.8 54.7 65.1 73.4 75.2 86.3 87.8 91.0 95.3	(4.103) (2.052) (1.937) 1.639 1.594 1.745 (1.825) (1.801) (1.675)	61.28 ^d 61.28 ^d 122.4 94.32 82.76 65.19 42.70 37.99 34.24 32.93	(731000) (866 724 756 155 35.28 18.8 47.1 30.0 21.85	0.0838)[0.0899] 0.0708 0.169[0.179] 0.1249[0.1786] 0.534 1.848 $2.265 \le 0.8073[1.250]$ 1.141[1.428] 1.507[1.696]	20.39 23.65 4.224 77.36 90.18 85.24	[139.2] 1.112 1.128 [138] 1.024 [34.9] 	0.2 GH2 LH2 LHe LH	LU CH4 Be	B C AI
	Ne Al	10 13	20.1797 26.981539	0.49555 0.48181	66.1 70.6	96.6 106.4	(1.724) 1.615	$28.94 \\ 24.01$	24.0 8.9	1.204[0.9005] 2.70	27.09	1.092[67.1]	Source: NAS	SA, Glenr tv Manua	n Research I
Source Exclored minor	$\frac{d\varepsilon_n}{ds} = -\frac{1}{2}$ Ioniza	$\frac{1}{\beta^2} \frac{dI}{dx}$ $\frac{dE}{dx}$ tion c	$\frac{E_{\mu}}{ls} \frac{\varepsilon_n}{E_{\mu}} + \frac{\varepsilon_n}{\varepsilon_{\mu}} +$	$\frac{1}{\beta^{3}} \frac{\beta_{\perp}}{2E_{j}}$ $\frac{1}{L_{R}} \alpha$ Beth	$\frac{0.014}{m_{\mu}L_{R}}$	a potential Access				If Material Com GH2 11 Aluminitiesand i Austenitic staint 16, nickel jonet as 321, 347) Carbon Meels ss, Copper Yind its a bronze, and cop Gray, divide, or Low-all/isostech Nickel joid is	pataSilityr ervice Maligrial SL itYaHoys Yo less steels w , Wo4, 304K/ No N illögs (sucYo pper-nickel) r Oist iron N s No N	fanyldydrogen S H2 es Aluminu ith > SomAustaaiti easos Stressaithabi@ tempêduluêf o Too Gantheria as, bra: Copper a bronze, jo Not Şienyitha jo Too brittelifo o Too brittelifo	Service Summary of RemiMaterial m and its alloys its stainkies etoclewith 3178 aughets, pilott alothay 308, 31 7) 2 tochyogenic service. 0 ind its alloys (such as, bras and copper-nickel) differ, hydrogdits service. 4 morkendenic service. 4 workendenic service. 4 workendenic service. 4	idvlageridlot/oid <u>GHB</u> em Yes wine make marten téssed aboVegriek mperature, oo brittle friesryo; s, Yes ot permitteNdor h so brittle Riesryo; recentible Bidwirk	repatability for H Service arkH2 SLH2 Yes Yes sitic conversion if S provident of S provident of S provident of S yes Yes yelNo servident of T provident of S provident of S yelNo servident of S provident of S
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7	LH ₂ operating temperature			01	2	к MPa			₄∖_₄	9 % Ni) TitaninWand its	Mhave V	temperatures	and its allows	mperatures. Nav	ti Yas Yau
	LH2 density H2 boiling point at 0.12 MPa			74.2	28 j.	$\frac{kg/m^3}{K}$		┨┢		³⁰ Asbesto¥ t8pres	g liste d with	EifforAvoidsbestbe	cimpeogricatedinoightfileflon [®]	void use b¥ænse (izard.	olYcascinogYuic /
	H ₂ freezing point at 0.12 MPa			14	1	K		φ	~	Chloroprese rub Daeron¥es	bbio(Neoph No N	nte ⁸⁸) Too Ghittio fu lo Too Dhittio f o	еверифрас (bicopeene®) [o r cryogenic service.] [o	oo brittle féice ryog 10 brittle féice ryog	geiño servicko — T geiño servicko — T
	LH ₂ viscosity 3 LH ₂ specific heat 7				$\frac{10}{10}$	⁻⁶ Pa-s				Fluorochibon ru Mylar [®] Yes	bbbr (Vitob) No N	 Too Huitthedia Too Mijittie fo 	rhenyogibhier (Aritice. [®]) (c r cryogenie service. (c	oo brittle féice ryog oo brittle féice ryog	ge iño servickio – T geiño servickio – T
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	LH ₂ thermal c	onducti	bility	97	7 m'	W/m-K				¹ Polychlöiotriflu Polychlöiotriflu	oféthylene)(Kel-F [®] Polychlo Bolytotro	rotrifluorethylene (Kel-F [®]) fluorethylene (Teflor [®])	Yes	Yes Yes Yes Yes
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Disadvantages of the LH₂ choice

→ Stringent safety requirements for mech. & elec. designs (25 liters of LH_2 released into the air with a 10% yield = 4 kg of TNT)

- \rightarrow Need Cryogenic plant to supply LHe flow and subcooled LH₂ to 17 K
- \rightarrow None standard thin Aluminum windows to contain LH₂
- On-going effort to comply with safety requirements

Cryogenic Properties of Various Gasses

• Supported by photogrammetry: Innovative none intrusive technique

KOTUBAL LEGULATION								
Parameter	He	N_2	H_2	CH ₄	C_3H_8			
Triple point temperature $T_t(K)$	2.177^	63.15	13.81*	90.69	91.46			
Heat of fusion (a) T _t (J g ⁻¹)	-NA-	25.3	59.5*	58.41	79.9 7			
Boiling temp. T b @ 1 bar (K)	4.222	77.35	20.28*	111.67	230.46			
Liquid density $\rho_1 @ T_b (kg m^{-3})$	124.9	807	70.8*	422.4	585.3			
Gas density ρ_g @ T_b (kg m ⁻³)	16.89	4.622	1.339*	1.816	2.497			
Gas to liquid volume ratio at T $_{\rm b}$	7.395	175.6	52.87*	232.6	298.0			
Gas V_{293} to Liquid V_b Ratio	699.4	645.6	792.9	591.4	379.0			
Gas C _p @ T _b (J g ⁻¹ K ⁻¹)	9.144	1.341	12.24*	2.218	1.642			
Heat of vaporization @ T $_{b}$ (J g ⁻¹)	20.7	198.8	445*	510.8	424.8			
Heat flux for $\Delta T=300$ -T _b (kWm ⁻²)	~200	~27	~93	~47	-NA-			
Broken vacuum heat flux (kWm ⁻²)	~35	~1.6	~19	~0.31	-NA-			
Critical temperature T _c (K)	5.195	126.2	32.98	190.6	368.8			
Critical pressure P _c (MPa)	0.228	3.39	1.29	4.59	4.36			

[^] The lambda point temperature for helium. For helium liquid, gas, and solid can't coexist.
 * This data is for para hydrogen. Ortho hydrogen changes to para hydrogen at <100 K.

 CEC 30 August 2005
 Paper C1-J-03
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 M. Green: "Hydrogen Safety Issues compared to Safety Issues with Methane and Propane", CEC'05

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Gas H2: NASA, Glenn Research Properties of gaseous (normal)

527.7° R
101.325 abs
83.7 g/m ³
0.0119 m ³ /g
Cp= 14.33 J/g-k
Cv= 10.12 J/g-k
1294 m/sec
Low= 119.93 kJ/g
High= 141.86 kJ/g
Upper= 75 % volume
Upper= 95 % volume
Upper= 59 % volume
Upper= 90 % volume
0.02 mJ
0.007 mJ

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Effort is on-going within two schemes

LH2: Convection scheme : Advantage of limit LH₂ inventory for MICE: energy deposition ~100W



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Review of the basic LH₂ Absorbers designs - Convection scheme

Numerical simulation (K.Cassel/IIT)

- Natural transverse convection
- Self regulating

Average Convective Heat Transfer Coefficient vs. Rayleigh Number



<u>Proposal:</u> →Heat transfer area=0.2m², →mdot=28 g/s

Design by S. Ishimoto [see MAC's talk]

Orders of magnitude: {L=0.3m; R=20cm;Q=150 W} → Ra=7.25e13

Convection heat transfer coefficients: {Thein/out=14K/15 K;T_ LH₂ =18.5 K} \rightarrow h_LHe=1580 W/m²K \rightarrow h_ LH₂ =210 W/m²K



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Review of the basic LH₂ Absorbers designs - Convection scheme

Experimental test (FNAL/KEK)6.2-liter of LH2LNe test @ KEK; LH2 tests @ KEK run 1 & 2





KEK Run1: some Lessons learned
Safety review process and design requirements
Operating procedures and flow sensitive
Cryo-system and LH₂ absorber stabilities
Instrumentation precision (TT, FT)
Data interpretation: heater element influence
Numerical simulations to valid test <-> prototype ?



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[FNAL Internal report-11/05 "Reflections on Initial MTA Tests ", B.Norris, C.Darve, L.Pei]

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Effort is on-going within two schemes

LH2: Convection scheme : Advantage of limit LH₂ inventory for MICE: energy deposition ~100W

[IIT Internal Report-Cassel et al. 2001] [NUFACT'01 – S. Ishimoto et al.] [PAC'05 – MA.Cummings and S.Ishimoto] (Cryocooler ~ 20 W) LH2: forced-flow schement Advantage of maximum energy removed for MuCool (energy deposition [Adv. Cryo. Engr.,2004–C. Darve [ICEC19, 2002 –C. Darve et al.)



External cooling loop for the LH₂ forced-flow absorber

- → Cryo-loop design with minimum of 25-liter
- → Nozzle design and flow distribution to validate by a prototype test

→ Since 2001 few numerical simulations have permitted to optimize the windows geometry while complying with safety requirements - maximum allowable working stress in the aluminum ($P_LH_2 = 1.2$ atm).

- → minimize thickness => to reduce multiple scattering
- → modify windows curve > to improve heat transfer



[Design by E. Black, W.Lau, S.Yang]

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Review of the basic LH₂ Absorbers designs – Forced-flow scheme



MuCool ~300W

- Numerical simulation by Oxford Univ.:
 - Holding/Lau using Algor 2D (2002)



•Yang/Lau using CFX 3D (2003)

Model	Inlet velocity	Initial Temp.	Heater	Reynolds Number	Temp. increase	Max. Pressure
8+12	2 m/s	17K	150W	4.2547E5	0.16K	279 Pa
11+15	0.5m/s	17K	150W	9.0158E4	1K	13.7 Pa
	2 m/s	17K	150W	3.49E5	0.38K	217 Pa

	config.\ velocity	0.5m/s	1m/s	2m/s	3.3m/s	4.4m/s
	11+15	8.6K	5K	0.96K	-	-
With Chamber	13+19	-	5.3K	2.47K	-	1.1K
	15+19	6.1K	3K	1.8K	0.86K	0.63K
Without Chamber	11+15	-	-	0.45K	-	-
	13+19	-	-	0.93K	-	-
	15+19	-	-	0.39K	-	-







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Velocity distribution:

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Pressure distribution:



These results are still waiting to be validated by experiment !!





Ex: 11+15 config _without channels _ inlet velocity at 2m/s, heater at 300W

Influence of flow symmetry and number of active nozzles.

-> Configuration with 15 inlet nozzles and 19 outlet nozzles; velocity 2 m/s at inner chamber inlet

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25-liter of LH₂

27 g/s of helium @ 14 K, 1.2 atm



- -Thermal analysis
- Hydraulic analysis (w/ Oxford Univ.)
- Process and Instrumentation Diagram
- Venting considered
- Oxygen Deficiency Hazard

Cryo-system development

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Design philosophy for a hostile environment

Electrical components located in the experimental hall must be rated Class 1, Div. 2 Gp B or considered as intrinsically safe: e.g. He guard, N₂ purge, specific equipments, etc...

Venting system designed to account for H₂ gas released from thermal expansion

Building equipment: flammable gas, Oxygen Deficiency Hazard system, ventilation, audio and visual alarm

>Operating Procedures , interlock, Safety PLC, Personnel safety training

Additional instrumentation and materials requirements:

- Operate at cryogenic temperatures (physical properties)
- High magnetic field (absorber sits inside the 4 Tesla SC solenoid magnet)
- Radiation-hard (able to withstand more than 300 mS/hr)

Design guidelines:

Safety codes: ASME, NEC, Fermilab ES&H Manual for pressure vessel and hydrogen target, e.g. "Guidelines for the Design, Fabrication, Testing, Installation and Operation of LH₂ Targets – 20 May 1997" by Del Allspach et al.

NASA: "SAFETY STANDARD FOR HYDROGEN AND HYDROGEN SYSTEMS: Guidelines for Hydrogen System Design, Materials Selection, Operations, Storage, and Transportation

LH₂ expertise from FNAL, SAMPLE, E158, G0...

LHC experiences: Cernox use

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Implementing SAMPLE cryo-loop

Permit to save fabrication cost (heat exchanger, operational LH₂ pump and motor) to help get the forced-flow system online, but the current engineering design must reflect these modifications







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Cryo-systems to support LH₂ absorber tests at MTA [see B.Norris's-talk]

- MTA facility equipped with Hydrogen test capacity
- MTA building: practicality of cryogenic cooling in hazardous environment
- Safety PLC (QUADlog), ODH, fan, vent, etc...

→ First MTA helium distribution using helium and nitrogen Dewars to operate KEK convective-type LH_2 absorber test and the SC solenoid magnet

→ Future Helium facility (compressor, refrigerator and transfer line) operating in switch-over mode [COOL'05]

- 4 K mode to SC MTA solehoid magnet and other cryo-components.
- > 14 K mode to forced-flow, convective type LH_2 absorbers

Use Tevatron satellite components \rightarrow Expected liquefaction rate is 78 liter/hr

→ Expander test has been performed to ensure the supply of 27 g/s of helium at 14 K, 15 psig (0.2 MPa) required for removing 300 W of heat deposition from the forced-flow absorber cryo-loop (Delta-F of 1 K)

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[COOL'05 – "Cryogenics for the MuCool Test Area (MTA)" C.Darve, B.Norris, L.Pei]



What exist and what we need ?

What exist for future LH₂ absorber tests?

→ Numerical models to simulate test results (convective and forced-flow schemes)

- More precise numerical model to account for experimental imperfections?
- Adequate instrumentation type, number and position to correlate data?

→ Expertise from KEK run : value of designing safety into the system from scratch ... Cryo-system design & fabrication, KEK cryostat analysis & test and finally operation

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- →Towards a continuous supply of LHe [See B.Norris's talk]
 - ✓ Installation of compressors
 - \checkmark Completion of the warm piping for the compressor room

What is needed for future H_2 absorber tests?

- > Towards a continuous supply of LHe
 - Complete design, install and test transfer lines
 - Complete design, install and test the refrigeration system
 - System review documentation and finally commissioning...



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Completed LH₂ **cryo-loop** Satisfy the safety requirements for the thin window design: pressure and design by integrating SAMPLE rupture tests, material certification, equipments Fermi National Accelerator Laboratory safety review document, etc... Safety Review approval Safety Review approval Fabricate and assemble parts Jryogenic Department Test the LH2 forced-flow absorber at MTA (in SC solenoid magnet) Complete the helium cryo-plant: Safety Review approval **Refrigerator room:** Compressors room: Complete the compressors refurbishing **Complete design and** manufacture as well as control&monitoring system (mechanical + electrical)

How things might proceed – Forced-flow LH₂ absorber test

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Concluding comments

On-going cryogenic efforts for the LH₂ absorbers CFD models exist and need to be validated by experiment - Forced-flow type: 3D by Oxford (2003)

- Convective-type: 2D by IT (2001)

Safety review in-progress for the KEK convective-type run2

LH₂ cryo-loop conceptual design available since 2003 Design modification requested to implement SAMPLE heritage Limited engineering resource available from FNAL/Cryo

> First LH₂ convective-type H_2 absorber tested at MTA (summer 2004). Limited energy deposition. Lesson learned..

> Hydrogen facility was successfully designed, built and commissioned at Fermilab to support LH₂ convection absorber during summer 2004

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